

## IONIZATION CONSTANTS OF ACIDS AND BASES

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These  $pK'_a$  values were taken from the original literature and from several extensive compilations of such data, of which the most important are

Albert, *Ionization Constants of Acids and Bases*, Methuen, London, 1962.

Bell, *The Proton in Chemistry*, 2nd ed., Cornell, Ithaca, New York, 1973.

Brown, McDaniel, and Häfliger, in Braude and Nachod, *Determination of Organic Structures by Physical Methods*, Academic Press, New York, 1955.

Kortum, Vogel, and Andrussov, *Dissociation Constants of Organic Acids in Aqueous Solution*, Butterworths, London, 1961.

Perrin, *Dissociation Constants of Organic Bases in Aqueous Solution*, Butterworths, London, 1965.

Yukawa, Ed., *Handbook of Organic Structural Analysis*, Benjamin, New York, 1965.

A particularly valuable source of dissociation constants obtained under a variety of experimental conditions is provided by Sillen L. G. and Martell, A. E., Eds., *Stability Constants*, Special Publications No. 17 and 25, Chemical Society, London, 1964 and 1971. This compilation also lists association constants of metals for a variety of inorganic and organic ligands.

The compounds selected were those which were thought most likely to be useful to biochemists and chemists and these compilations should be consulted for information on compounds which are not included here.

All values are reported as  $pK'_a = -\log K'_a = 14 - pK'_b$ .  $K'_a$  is the ionization constant

$$\frac{[H^+][A^-]}{[HA]} \text{ or } \frac{[H^+][B]}{[HB^+]} \text{ or } \frac{[A^{n-}][H^+]}{[HA^n]}$$

Temperatures are not indicated because variations of  $pK'_a$  with temperature are generally smaller than the variations of the data from different sources for other reasons, but most of the data were obtained at or near 25°. Ionization constants which are reported as thermodynamic values at 25° are indicated with an asterisk, \*, but some of these may only represent values measured at low ionic strength.

These  $pK'_a$  values and a measured pH should not be used to obtain an *exact* measure of the ratio of acid to base in a given solution. Ionic strength and specific salt effects, as well as possible errors in the reported  $pK'_a$  values, are likely to make such estimates inaccurate. It should be kept in mind that the effect of increasing ionic strength is generally to decrease the apparent  $pK'_a$  of neutral and anionic acids and to increase the  $pK'_a$  of cationic acids. These effects are particularly large for polyanions, such as phosphates.

There is some intentional redundancy in the tables to facilitate the location of listings for compounds that might be listed in several sections. The  $pK'_a$  values for amines refer to the ionization of the conjugate acids of the amines except for a few nitrogen acids, which undergo an acidic ionization.

The pH of a solution at a given ionic strength and temperature is given by

$$pH = pK'_a + \log \frac{[\text{base}]}{[\text{acid}]}$$

in which the  $pK'_a$  is measured under the same experimental conditions. The following relationships are useful to have readily available to estimate the ratio of acid to base at a given pH or to estimate the buffer ration of acid required to give a given pH; the compiler keeps a copy of these numbers on his desk.

Fraction base or acid	pH	Fraction base or acid	pH
5% or 95%	$pK'_a \pm 1.25$	30% or 70%	$pK'_a \pm 0.37$
10% or 90%	$pK'_a \pm 0.95$	35% or 65%	$pK'_a \pm 0.27$
15% or 85%	$pK'_a \pm 0.75$	40% or 60%	$pK'_a \pm 0.18$
20% or 80%	$pK'_a \pm 0.60$	45% or 55%	$pK'_a \pm 0.09$
25% or 75%	$pK'_a \pm 0.48$	50% or 50%	$pK'_a \pm 0$

## AMINO ACIDS

Compound	pK <sub>a</sub>	Reference	Compound	pK <sub>a</sub>	Reference
Alanine	2.34, 9.69	6	$\alpha$ -Aminotricarballylic acid	2.10, 3.60,	99
<i>N</i> -Acetyl-	3.72	97		4.60, 9.82	
Amide	8.02*	99	$\alpha$ -Aminovaleric acid	4.20	130
3-(2-Aminoethylthio)-	8.28, 9.30	99	2-Anilinoethylsulphonic acid	3.80	99
Carbamyl-	3.89	99	Arginine	12.48, 2.17,	6
<i>N</i> -Ethyl-	2.22, 10.22	99		9.04	
<i>N</i> -Methyl-	2.22, 10.19	99	Argininosuccinic acid	> 12, 1.62,	
<i>N</i> -n-Propyl-	2.21, 10.19	99		9.58, 2.70,	
$\beta$ -(2-Pyridyl)-	1.37, 4.02, 9.22	99		4.26	
$\beta$ -(3-Pyridyl)-	1.77, 4.64, 9.10	99	Asparagine	2.02, 8.8	6
$\beta$ -(4-Pyridyl)-	4.85	99	$\alpha$ -Hydroxy-	2.28, 7.20	99
$\beta$ -Alanine	3.60, 10.19	6	$\beta$ -Hydroxy-	2.09, 8.29	99
<i>N</i> -acetyl-	4.44	129	Aspartic acid	2.09, 3.86, 9.82	99
Carbamyl-	4.49	129	Diamide	7.00	99
Allothreonine	2.11, 9.01	99	Hydroxy-	1.91, 3.51, 9.11	99
<i>O</i> -Methyl-	1.92, 8.90	99	Azaserine	8.55	101
$\gamma$ -Aminoacetoacetic acid	2.9, 8.3	99			99
$\alpha$ -Aminoadipic acid	2.14, 4.21	101			99
2-Aminobenzoic acid	2.19, 4.95	99			99
<i>N</i> , <i>N</i> -Dimethyl-	1.4, 8.49	99	$\gamma$ -Butyrobetaine	3.94	99
3-Hydroxy-	5.19, 10.12	99	Canaline	2.40-, 3.70,	99
<i>N</i> -Methyl-	1.97, 5.34	99		9.20	
3-Aminobenzoic acid	3.29, 5.10	99	Canavanine	2.50, 6.60, 9.25	99
4-Aminobenzoic acid	2.50, 4.87	99	L-Citrulline	2.43, 9.41	99
4-Aminobutylphosphonic acid	2.55, 7.55, 10.9	99	Creatine	2.67, 11.02	6
4-Aminobutylsulphonic acid	10.65	99	Creatinine	4.84, 9.2	6
$\alpha$ -Aminobutyric acid	2.55, 9.60	6	Cycloserine	4.4, 7.4	101
Carbamoyl- $\alpha$ -amino- <i>n</i> -butyric	3.89	129	Cysteine	10.78, 1.71,	6
$\gamma$ -Aminobutyric acid	4.23, 10.43	6		8.33	
Carbamyl-	4.68	129	Ethyl ester	6.69, 9.17	99
2-Aminobutyric acid	2.27, 9.68	99	Methyl ester	6.56, 8.99	99
$\alpha$ -Amino- <i>n</i> -caproic acid	2.33	129	<i>S</i> -Ethyl-	1.94, 8.69	99
$\epsilon$ -Aminocaproic acid	4.37	129	<i>S</i> -Methyl-	8.75	99
10-Aminodecylphosphonic acid	8.0, 11.25	99	Cystine	1.65, 7.85	6
10-Aminodecylsulphonic acid	11.35	99	L-Cystine diamide	5.93, 6.90	99
10-Amino- <i>n</i> -dodecanoic acid	4.648	99	2,4-Diaminobutyric acid	1.85, 8.28,	99
Aminoethylphosphoric acid	2.45, 7.0, 10.8	99		10.50	
2-Aminoethylsulphonic acid	8.95	99	2,3-Diaminopropionic acid	1.23, 6.73, 9.56	99
$\omega$ -Aminoheptanoic acid	4.50	136	2,7-Diaminosuberlic acid	1.84, 2.64,	99
6-Aminohexanoic acid	4.37, 10.81	99		9.23, 9.89	
$\alpha$ -Aminoisobutyric acid	2.36, 10.21	6	3-Dimethylaminopropionic acid	9.85	99
Carbamyl-	4.46	129	Formamidinoglutamic acid	2.7, 4.4, 11.3	99
$\alpha$ -Aminoisocaproic acid	2.33	129	Formamidinoacetic acid	2.6, 11.5	99
$\alpha$ -Aminoisovaleric acid	2.29	129	Glutamic acid	2.19, 4.25, 9.67	6
$\delta$ -Aminolaevulinic acid	4.05, 8.90	99	Diethyl ester	7.04	99
Aminomethylphosphonic acid	2.35, 5.9	99	$\gamma$ -Monobenzyl ester	2.17, 9.00	99
Aminomethylsulphonic acid	5.75	99	$\alpha$ -Monoethyl ester	3.85, 7.84	99
$\alpha$ -Amino- $\beta$ -methyl- <i>n</i> -valeric	2.32	129	$\gamma$ -Monoethyl ester	2.15, 9.19	99
acid			Glutamine	2.17, 9.13	6
1-Aminonaphthalene-2-	1.71	99	Glycine	2.34, 9.6	6
sulphonic acid			<i>N</i> -Acetyl-	3.67	99
2-Aminonaphthalene-1-	2.35	99	<i>N,N</i> -bis(2-Hydroxyethyl)-	2.50, 8.11	99
sulphonic acid			<i>N</i> -n-Butyl-	2.35, 10.25	99
3-Amino-1-naphthoic acid	2.61, 4.39	99	Carbamyl-	3.88*	97
4-Aminopentanoic acid	3.97, 10.46	99	Chloroacetyl-	3.38*	97
5-Aminopentylsulphonic acid	10.95	99	<i>N,N</i> -Diethyl-	2.04, 10.47	99
4-Aminophenylacetic acid	3.60, 5.26	99	Dihydroxyethyl-	8.08*	97
2-Aminophenylarsonic acid	3.77, 8.66	99	<i>N,N</i> -Dimethyl-	2.08-, 9.80	99
2-Aminophenylboric acid	4.53, 9.31	99	<i>N</i> -Ethyl-	2.34*, 10.23	99
$\beta$ -Aminopropionic acid	3.55*, 10.23*	97	Ethyl ester	7.83	99
4-Aminosalicylic acid	1.78, 3.63	99	Formyl-	3.43*	97
			<i>N</i> -Isobutyl-	2.35, 10.12	99
			Methyl ester	7.73	99

\* Thermodynamic value.

## PEPTIDES

Compound	pK <sub>a</sub>	Reference	Compound	pK <sub>a</sub>	Reference
Ala-Ala-(LD)	3.12, 8.30	27	Gly-Ala-Ala (LD)	3.30, 8.17	27
Ala-Ala-(LL)	3.30, 8.14	27	Gly-Ala-Ala (LL)	3.38, 8.10	27
Ala-Ala-Ala-(3D)	3.39, 8.06	27	Gly-Ala-Ala-Gly	3.30, 7.93	99
Ala-Ala-Ala-(DLL)	3.37, 8.06	27	Gly-Asp	2.81, 4.45, 8.60	99
Ala-Ala-Ala-Ala-(DLLL)	3.42, 7.99	27	Gly-asparagine	2.82, 7.20	99
Ala-Ala-Ala-(3L)	3.39, 8.03	27	Gly-Gly	3.06, 8.13	6
Ala-Ala-Ala-Ala-(4L)	3.42, 7.94	27	Gly-Gly-cystine	2.71, 7.94	99
Ala-Ala-Ala-(LDL)	3.31, 8.13	27	Gly-Gly-Gly	3.26, 7.91	23
Ala-Ala-Ala-Ala-(LDLL)	3.22, 7.99	27	Gly-His	6.79, 8.20	99
Ala-Ala-Ala-(LLD)	3.37, 8.05	27	Gly-Leu	3.10, 8.41	99
Ala-Ala-Ala-Ala-(LLDL)	3.24, 7.93	27	Gly-Pro	2.81, 8.65	99
Ala-Gly	3.16, 8.24	27	Gly-sarcosine	2.98, 8.57	99
Ala-Gly-Gly	3.19, 8.15	99	Gly-Ser	2.92, 8.10	99
Ala-Lys-Ala-(3L)	3.15, 7.65, 10.30	27	Gly-Ser-Gly	3.23, 7.99	99
Ala-Lys-Ala-(LDL)	3.33, 7.97, 10.36	27	Gly-Trp	8.06	99
Ala-Lys-Ala-(LDLL)	3.32, 8.01, 10.37	27	Gly-Tyr	2.93, 8.45, 10.49	99
Ala-Lys-Ala-(LLD)	3.29, 7.84, 10.49	27	Gly-Val	3.15, 8.18	99
Ala-Lys-Ala-Ala-(4L)	3.58, 8.01, 10.58	27	His-Gly	2.36, 6.27, 8.57	99
Ala-Lys-Ala-Ala-Ala-(5L)	3.53, 7.75, 10.35	27	His-His	5.54, 6.80, 7.82	99
Ala-Lys-Ala-Ala-Ala-(LDLL)	3.30, 7.85, 10.29	27	Leu-asparagine	2.83, 8.23	99
$\beta$ -Ala-1-methylhistidine	2.64, 7.04, 9.49	99	Leu-Tyr	2.87, 8.36, 10.28	99
Ala-Pro	3.04, 8.38	99	Lys-Ala-(LD)	3.00, 7.74, 10.63	27
$\beta$ -Ala-Bis	2.73, 6.87, 9.73	99	Lys-Ala-(LL)	3.22, 7.62, 10.70	27
Anserine	7.0, 2.65, 9.5	6	Lys-Glu	2.98, 4.47, 8.45, 11.30	99
Asparaginy-Gly	2.90, 7.25	99	Lys-Lys-(LD)	2.85, 7.53, 9.92, 10.98	27
Asp-Asp	2.70, 3.40, 4.70, 8.26	99	Lys-Lys-(LL)	3.01, 7.53, 10.05, 11.01	27
$\alpha$ -Aspartyl-histidine	2.45, 3.02, 6.82, 7.98	99	Lys-Lys-Lys-(3L)	3.08, 7.34, 9.80, 10.54, 11.32	27
$\beta$ -Aspartyl-histidine	1.93, 2.95, 6.93, 8.72	99	Lys-Lys-Lys-(LDD)	2.94, 7.14, 9.60, 10.38, 11.09	27
Asp-Gly	2.10, 4.53, 9.07	99	Lys-Lys-Lys-(LDL)	2.91, 7.29, 9.79, 10.54, 11.42	27
Asp-Tyr	2.13, 3.57, 8.92, 10.23	99	Met-Met	2.22, 9.27	99
Carnosine	6.83, 9.51	6	Methyl-Leu-Gly	3.29, 7.82	99
Cys-Cys	2.65, 7.27, 9.35, 10.85	99	Phe-Ala-Arg	2.60, 7.54, 12.43	99
Cys-Gly-Gly	3.13, 6.36, 6.95	99	Phe-Gly	3.13, 7.62	99
Cys-Gly-Gly-Gly-Gly	3.21, 6.01, 6.87	99	Phenylalanylglycine amide	6.72	99
L-Cystinylcystine	1.87, 2.94, 6.53, 7.66	99	Pro-Gly	3.19, 8.97	99
<i>N,N</i> -Dimethylglycyl-glycine	3.11, 8.09	99	Sarcosyl-Gly	3.14, 8.66	99
<i>N,N</i> -Dimethyl-leucyl-glycine	7.78	99	Sarcosyl-Leu	3.15, 8.67	99
Glutamyl-glutamic acid	3.14, 4.38, 7.62	99	Sarcosylsarcosine	2.89, 9.18	99
Glutamyl-glycine	3.15, 7.52	99	Ser-Gly	3.10, 7.33	99
Glutathione	3.59, 8.75, 9.65	77	Ser-Leu	3.08, 7.45	99
Glutathione, oxidized	3.15, 4.03, 8.57, 9.54	77	Tyr-Tyr	3.52, 7.68, 9.80, 10.26	99
Gly-Ala (L), (D)	3.17, 8.23	27	Val-Gly	3.23, 8.00	99

## THIOLS

Compound	pK <sub>a</sub>	Reference	Compound	pK <sub>a</sub>	Reference		
<i>N</i> -Acetylcysteine	9.52	112	<i>o</i> -Mercaptophenylacetic acid	4.28, 7.67	59		
<i>N</i> -Acetyl- $\beta$ -mercaptoisoleucine	10.30	112	2-Mercaptopropionic acid	4.32, 10.30	153		
<i>N</i> -Acetylpenicillamine	9.90	112	Methyl cysteine	6.5, (7.5)	81		
<i>O</i> -Aminothiophenol	6.59	81	Methyl- $[\beta$ -diethylaminoethyl]-sulfide	9.8	5		
<i>p</i> -Chlorothiophenol	7.50	81	Methyl thioglycolate	7.8	23		
Cysteine	1.8, 8.3, 10.8	23	<i>p</i> -Nitrobenzenethiol	5.1	58		
Cysteine ethyl ester	6.53, 9.05	112	Penicillamine	7.90, 10.42	112		
Cysteinylcysteine	2.65, 7.27, 9.35, 10.85	23	Thiocyanic acid	-1.84	104		
1-Diethylamino-butane-(4)	10.1	5	Thioglycolic acid	3.67, 10.31	23		
1-Diethylamino-hexane-(6)	10.1	5	Thiophenol	7.8, 6.52	59, 81, 82		
1-Diethylamino-propane-(3)	8.0, 10.5	5	Pentafluoro-	2.68	155		
<i>N</i> -Diethyl-cysteamine	7.8, 10.75	5	<i>p</i> -Me-	6.82	157		
<i>N</i> -Dimethyl-cysteamine	7.95, 10.7	5	<i>p</i> -OMe-	6.77	157		
<i>N</i> -Dipropyl-cysteamine	8.00, 10.8	5	<i>m</i> -Me-	6.66	157		
Ethyl mercaptan	10.50	81	<i>m</i> -OMe-	6.38	157		
Glutathione	2.12, 3.59, 8.75, 9.65	23	<i>p</i> -Cl-	6.13	157		
DL-Homocysteine	8.70, 10.46	112	<i>p</i> -Br-	6.02	157		
2-Mercaptoethanesulfonate	7.53 (9.1)	81	<i>m</i> -Cl-	5.78	157		
Mercaptoethanol	9.5	23	<i>p</i> -COMe-	5.33	157		
Mercaptoethylamine	8.6, 10.75	23	<i>m</i> -NO <sub>2</sub> -	5.24	157		
<i>N</i> - $\beta$ -Mercaptoethylmorpholine	6.65, 9.8	5	<i>p</i> -NO <sub>2</sub> -	4.71, 4.50	157		
<i>N</i> - $\beta$ -Mercaptoethylpiperidine	7.95, 11.05	5	<i>L</i> -Thio-D-sorbitol	9.35	81		
$\beta$ -Mercaptoisoleucine	8.10, 10.6	112	<i>N</i> -Trimethyl cysteine	8.6	23		
X =	-H	-S <sup>-</sup>	-SH	X =	-H	-S <sup>-</sup>	-SH
X(CH <sub>2</sub> ) <sub>2</sub> SH	12.0	13.96	10.75	X(CH <sub>2</sub> ) <sub>3</sub> SH	—	13.24	11.14
X(CH <sub>2</sub> ) <sub>4</sub> SH	12.4	13.25	11.50	X(CH <sub>2</sub> ) <sub>5</sub> SH	—	13.27	11.82
Compound	pK <sub>a</sub>	Reference	Compound	pK <sub>a</sub>	Reference		
Mercaptans, RSH			<i>t</i> -C <sub>4</sub> H <sub>9</sub> -	11.05	82		
R			(CH <sub>3</sub> ) <sub>2</sub> CH-	10.86*	103		
C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> -	9.43	82	(CH <sub>3</sub> ) <sub>3</sub> C-	11.22*	103		
HOCH <sub>2</sub> CH(OH)CH <sub>2</sub> -	9.51	82	HOCH <sub>2</sub> CH <sub>2</sub> -	9.72	103		
CH <sub>2</sub> =CHCH <sub>2</sub> -	9.96	82	CH <sub>3</sub> CONHCH <sub>2</sub> CH <sub>2</sub> -	9.92	103		
<i>n</i> -C <sub>4</sub> H <sub>9</sub> -	10.66	82	-OCOCH <sub>2</sub> -	10.68*	103		
<i>t</i> -C <sub>3</sub> H <sub>7</sub> -	11.21	82	-OCOCH <sub>2</sub> CH <sub>2</sub> -	10.84*	103		
C <sub>2</sub> H <sub>5</sub> OCOCH <sub>2</sub> -	7.95	82	<i>o</i> -OCOC <sub>6</sub> H <sub>4</sub> -	8.88*	103		
C <sub>2</sub> H <sub>5</sub> OCH <sub>2</sub> CH <sub>2</sub> -	9.38	82	<i>p</i> -OCOC <sub>6</sub> H <sub>4</sub> -	5.80*	103		
HOCH <sub>2</sub> CH(OH)CH <sub>2</sub> -	9.66	82	CH <sub>3</sub> CO-	3.62*	103		
<i>n</i> -C <sub>3</sub> H <sub>7</sub> -	10.65	82					

\* Thermodynamic value.